A SIMPLE PHASE TRANSFER CATALYZED SYNTHESIS OF BENZYL AND HETARYLMETHYL SILACYCLOALKYL ETHERS FROM ACETATES Edgars Abele, Ramona Abele, Ilze Sleiksa, Edmunds Lukevics* Latvian Institute of Organic Synthesis, Aizkraukles 21, Riga, LV-1006, Latvia Fax: (371)7821038

Abstract: Simple and one-pot phase transfer catalytic method for the preparation of benzyl and hetaryl silacycloalkyl ethers from corresponding acetates in the system chloropropylsilane / KOH / KI / 18-crown-6 was elaborated. The corresponding benzyl and hetaryl ethers were obtained in 56 \rightarrow 78 % yield.

Introduction

Synthesis of O-trialkylsilylalkyl ethers under nucleophilic catalysis conditions are of interest in recent years. Application of Williamson ether synthesis to the preparation of benzyl trimethylsilylmethyl ether was unsuccessfull because of isomerization of (trimethylsilyl)methyl alcoholate to methoxytrimethylsilane (Brook rearrangement)^{1,2}. It was also known that in the reaction with halogenomethyltrimethylsilanes the alkoxide anions generated from alcohols with NaH in THF/DMF/ n-Bu₄NI system produced only cleavage products, i.e., methyl and trimetylsilyl ethers with no traces of the expected trimethylsilylmethyl ether³. Therefore, we have elaborated a synthetic procedure starting from acetates and using phase transfer catalysis.

Results and Discussion

We have found that silacycle containing benzyl and hetarylmethyl ethers 8-14 $^{4-5}$ can be successfully obtained from the corresponding acetates 1-7 6 in the phase transfer catalytic system 3-chloropropylsilane⁷ /solid KOH / solid KI/ toluene/18-crown-6. Obviously, the formation of ethers 8-14 proceeds through the corresponding alcohols not detectable by GC-MS due to the rapid changes. The remarkable observation was that the etherification reactions with bulky chloropropylsilanes in the absence of added KI were very slow and the yields of the products obtained were usually low.⁸ It suggests that this reaction proceeds via convertion of chloropropylsilane to the corresponding iodopropylsilane which is more reactive in the alkylation reaction.

$$\begin{array}{c} O \\ H \\ ArCH_2OCCH_3 \end{array} \xrightarrow[reflux]{} CI(CH_2)_3 S(\underline{)}_n \\ \hline KOH, KI, 18-crown-6, PhMe \\ reflux \end{array} ArCH_2O(CH_2)_3 S(\underline{)}_n \\ \hline Me \\ 8-14 \end{array}$$

Starting	Ar	Reaction	r		Bn (°C) at	Γ
o un timb		Reaction		1	Dp (C) at	
acetate		time, h	Product		10 mm Hg	Yield $(\%)^a$
1	Ph	3	8a	1	148-151	68
1	Ph	3	8b	2	168-170	67
2	2-furyl	5	9 a	1	134-136	64
2	2-furyl	6	9Ъ	2	145-146	63
3	2-thienyl	4	10 a	1	146-148	72
3	2-thienyl	4	10b	2	150-152	71
4	2-pyridyl	7	11a	1	149-150	56
5	6-methyl-2-pyridyl	9	12a	1	182-183	67
5	6-methyl-2-pyridyl	9	12b	2	190-191	69
6	3-pyridyl	7	13a	1	146-147	63
7	4-pyridyl	6	14a	1	135-136	78
7	4-pyridyl	6	14b	2	150-152	71

Table 1. Synthesis of aryl and hetaryl ethers 8-14

^aIsolated yield after purification

Thus, a simple and selective one-pot method for the preparation of benzyl and hetaryl silacycloalkyl ethers has been elaborated. The isolated yields of the products were 58 - 78 %.

References

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- (4) General procedure for synthesis of ethers 8-14. To a solution of acetate 1-7 (5 mmol), chloropropylsilane (5 mmol) and 18-crown-6 (0.066 g, 0.25 mmol) in 8 ml of dry toluene were added powdered KI (3.32 g, 20 mmol) and powdered KOH (1.12 g, 20 mmol). The mixture was stirred 3-9 h at reflux temperature and cooled up to room temperature. The solid was filtered off and the solvent was removed under reduced pressure. The crude residue was purified by distillation in vacuo or chromatographed on silica (eluent benzene) to afford the pure products 8-14.
- (5) Illustrative spectroscopic data:

8a: ¹H NMR (90 MHz, CDCl₃) δ 0.24 (s, 3H, CH₃), 0.73 (m. 6H, (CH₂)₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.70 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 3.62 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 4.67 (s, 2H, CCH₂), 7.49 (m, 5H, Ph). MS m/z (rel. intensity) 248 (M⁺, <1), 101 (80), 91 (100), 71 (10), 45 (10).

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8b: 'H NMR (90 MHz, CDCl₃) δ 0.22 (s, 3H, CH₃), 0.78 (m, 6H, (CH₂)₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 1.77 (m, 8H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂). 3.67 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 4.73 (s, 2H, CCH₂), 7.56 (m, 5H, Ph). MS m/z (rel. intensity) 262 (M⁺, <0.1), 160 (12), 113 (10), 101 (81), 91 (100), 43 (9).

9a: ¹H NMR (90 MHz, CDCl₃) δ 0.07 (s, 3H, CH₃), 0.53 (m, 6H, (CH₂)₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.60 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 3.44 (t, 2H, J = 7 Hz, CH₂CH₂CH₂Si), 4.47 (s, 2H, CCH₂), 6.33 (m, 2H, H-3 and H-4), 7.40 (m, 1H, H-5). MS m/z (rel. intensity) 238 (M⁺, < 1), 101 (38), 81 (100), 71 (12), 53 (15), 45 (11).

9b: ¹H NMR (90 MHz, CDCl₃) δ 0.29 (s, 3H, CH₃), 0.86 (m, 6H, CH₂CH₂CH₂C_{H₂}Si and silacycle CH₂(CH₂)₃CH₂), 1.64 (m, 8H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 3.76 (t, 2H, J = 7 Hz, CH₂CH₂CH₂Si), 4.73 (s, 2H, CCH₂), 6.62 (m, 2H, H-3 and H-4), 7.64 (m, 1H, H-5). MS m/z (rel. intensity) 252 (M⁺, < 1), 101 (21), 81 (100), 53 (10).

10a: ¹H NMR (90 MHz, CDCl₃) δ 0.18 (s, 3H, CH₃), 0.64 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.64 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 3.56 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 4.76 (s, 2H, CCH₂), 7.08 (m, 2H, H-4 and H-5), 7.35 (m, 1H, H-3). MS m/z (rel. intensity) 254 (M⁺, < 1), 102 (25), 97 (100), 71 (12), 53 (10), 45 (21).

10b: ¹H NMR (90 MHz, CDCl₃) δ 0.20 (s, 3H, CH₃), 0.76 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 1.76 (m, 8H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 3.64 (t, 2H, J = 6.6 Hz, CH₂(CH₂)₂Si), 4.87 (s, 2H, CCH₂), 7.15 (m, 2H, H-4 and H-5), 7.42 (m, 1H, H-3). MS m/z (rel. intensity) 268 (M⁺, < 1), 101 (21), 97 (100), 85 (19), 45 (15).

11a: ¹H NMR (90 MHz, CDCl₃) δ 0.20 (s, 3H, CH₃), 0.71 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.64 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 3.42 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 4.71 (s, 2H, CCH₂), 7.13 (m, 1H, H-5), 7.60 (m, 1H, H-3), 8.09 (m, 1H, H-4), 8.62 (m, 1H, H-6). MS m/z (rel. intensity) 249 (M⁺, 2), 220 (68), 192 (80), 178 (45), 165 (26), 152 (21), 136 (26), 101 (40), 93 (100), 65 (32), 43 (22).

12a: ¹H NMR (90 MHz, CDCl₃) δ 0.29 (s, 3H, SiCH₃), 0.78 (m, 6H, CH₂CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.76 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 2.67 (s, 3H, CH₃ in pyridine ring), 3.71 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 3.89 (s, 2H, CCH₂), 7.0-7.3 (m, 2H, H-3 and H-5), 7.71 (m, 1H, H-4). MS m/z (rel. intensity) 263 (M^{*}, 3), 234 (12), 206 (24), 192 (12), 150 (13), 122 (17), 107 (100), 79 (17).

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12b: ¹H NMR (90 MHz, CDCl₃) δ 0.26 (s, 3H, SiCH₃), 0.84 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 1.86 (m, 8H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 2.69 (s, 3H, CH₃ in pyridine ring), 3.76 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 3.93 (s, 2H, CCH₂), 7.0-7.3 (m, 2H, H-3 and H-5), 7.82 (m, 1H, H-4). MS m/z (rel. intensity) 277 (M⁺, 2), 234 (22), 206 (18), 192 (16), 179 (10), 122 (20), 107 (100), 85 (28), 59 (11), 43 (10). **13a**: ¹H NMR (90 MHz, CDCl₃) δ 0.18 (s, 3H, CH₃), 0.64 (6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.62 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 3.55 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 3.78 (s, 2H, CCH₂), 7.38 (m, 1H, H-5), 7.78 (m, 1H, H-4), 8.64 (m, 2H, H-2 and H-6). MS m/z (rel. intensity) 249 (M⁺, 2), 220 (12), 192 (17), 101 (100), 92 (81), 65 (26), 39 (17).

14a: H NMR (90 MHz, CDCl₃) δ 0.13 (s, 3H, CH₃), 0.64 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 1.62 (m, 6H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₂CH₂), 3.56 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 3.73 (s, 2H, CCH₂), 7.31 (m, 2H, H-3 and H-5), 8.60 (m, 2H, H-2 and H-6). MS m/z (rel. intensity) 249 (M⁺, 3), 220 30), 207 (14), 192 (39), 150 (13), 101 (100), 65 (23), 45 (19).

14b: ¹H NMR (90 MHz, CDCI₃) δ 0.20 (s, 3H, CH₃), 0.78 (m, 6H, CH₂CH₂CH₂C<u>H₂Si</u> and silacycle CH₂(CH₂)₃CH₂), 1.73 (m, 8H, CH₂CH₂CH₂Si and silacycle CH₂(CH₂)₃CH₂), 3.64 (t, 2H, J = 6.8 Hz, CH₂(CH₂)₂Si), 3.84 (s, 2H, CCH₂), 7.44 (m, 2H, H-3 and H-5), 8.71 (m, 2H, H-2 and H-6). MS m/z (rel. intensity) 263 (M⁺, 2), 220 (93), 207 (12), 192 (11), 178 (13), 150 (11), 113 (20), 101 (100), 92 (45), 65 (27), 43 (19).

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Received on October 30, 1998